Stefan W. Scheuplein^a, Klaus Harms^{b[+]}, Reinhard Brückner^{*a}, and Jean Suffert^{*c}

Institut für Organische Chemie der Julius-Maximilians-Universität^a, Am Hubland, W-8700 Würzburg, F.R.G.

Fachbereich Chemie der Philipps-Universität^b, Hans-Meerwein-Straße, W-3550 Marburg, F.R.G.

Laboratoire de Pharmacochimie Moléculaire, Centre de Neurochimie du CNRS UPR 421°, 5, rue Blaise Pascal, F-67084 Strasbourg Cedex, France

Received September 16, 1991

Key Words: C-C coupling / Dienediynes / Enol triflates / Neocarzinostatin / Palladium catalysis

The stereoselective synthesis of (Z)-dienediynes 2 related to the neocarzinostatin chromophore 1 is described. (Z)-Bis(enol triflate) 3, accessible stereoselectively from 2-formylcyclopentanone in 2 steps, serves as the key intermediate. It is subjected to consecutive Pd(0)-catalyzed couplings with two different alkynes and furnishes 2 via monocoupling products 8.

Neocarzinostatin is a highly potent antitumor chromoprotein from *Streptomyces*^[1]. Its biological activity stems from its chromophore 1^[2] which is readily transformed into a highly reactive diradical ready to attack — and thereby cleave — single- and double-strand DNA^[3]. Various synthetic approaches to 1 or simplified models thereof have been described during the past three years^[4].

Our own approach^[5] assumes that bicyclic dienediynes like 1 (or its analogs) would be accessible if monocyclic dienediynes 2 were at hand. In the synthesis of 2 endowing the semicyclic C=C bond with the proper (Z) configuration would be required. This seemed feasible if the correct stereochemistry was already established earlier, i.e. in the bis-(enol triflate) 3. From 3, we wanted to proceed to the dienediynes 2 by palladium(0)-mediated coupling reactions with alkynes. The application of these conditions should leave the geometry of the crucial double bond unchanged.

Recently, we have shown how this method was utilized in the synthesis of (Z)-dienediynes 7 derived from *one* kind of alkyne^[5]. At the same time we reported on an analogous synthesis of the corresponding (E)-dienediynes which was stereoselective, too^[6]. A later publication by Terashima et al.^[7] revealed closely related results.

Bistriflate 3 was obtained from formylcyclopentanone $4^{[4h,8]}$ by reaction of its lithium enolate 5 with triflic anhydride. Sulfonylation occurred selectively at the former formyl oxygen atom. The resulting monotriflate 6, due to its instability, was converted without purification into bistriflate 3. To this end, 6 was deprotonated with lithium hexamethyldisilazide at -65°C and sulfonylated once more. Using triflic anhydride for this purpose, we obtained bistriflate 3 in 29% yield in the two steps. N,N-bis(trifluoromethanesulfonyl)aniline [9], which we had employed previously, had secured 36% overall yield [5] but at distinctly higher costs.

The (Z) configuration of 3 did not emerge from NOE experiments: Irradiation of the endocyclic olefinic proton

^[+1] To whom inquiries concerning the X-ray structural analysis should be addressed.

 $[\delta = 6.23 \text{ (m}_c); \text{ distinguishable from 1'-H due to a 3.0-Hz}]$ coupling with two sp³-bound protons (cf. Experimental)] enhanced absorption by the neighboring 3-H₂ as expected $[\delta = 2.61 - 2.65 \text{ (m)}]$. However, irradiation of the semicyclic olefinic proton $[\delta = 6.54 \, (m_c)]$ had no effect on the integral of 4-H₂. This observation seemed to be in contradiction to

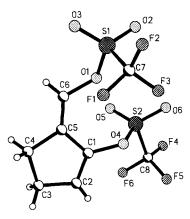
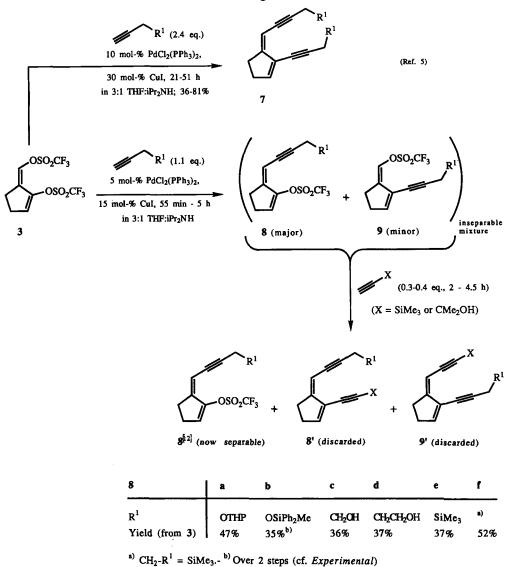


Figure 1. SCHAKAL plot of the solid-state structure of bistriflate $3^{[10]}$

the suggested (Z) configuration. On the other hand, we had prepared the bistriflate stereoisomer of 3 which also showed no NOE under similar conditions [6]. Recourse had therefore to be taken to low-temperature X-ray crystallography. As depicted in Figure 1, crystalline 3 possesses the desired (Z)double bond.

We have previously coupled 3 in the presence of catalytic $PdCl_2(PPh_3)_2$ and CuI with ≥ 2 equiv. of one alkyne. In THF/iPr2NH rather than in Cacchi's Et2NH/DMF[11] we obtained dienediynes 7 with two identical alkyne moieties [5].

In the present study we treated bistriflate 3 under otherwise similar conditions with a stoichiometric amount of the alkyne (in fact, 1.1 equiv. of alkyne was used but 0.1 equiv. was consumed by reduction of the added 0.05 equiv. of Pd^{II}). As a result, we obtained a mixture consisting only of the isomeric monocoupling products 8 (ca. 80-90%) and 9. This finding implies firstly that the bistriflate 3 undergoes coupling faster than any of the newly formed monotriflates. Secondly, it shows that the exocyclic triflate moiety of 3 couples more readily with alkynes than its endocyclic counterpart; this order would have been expected on steric grounds.



Unfortunately, we were unable to separate 8 from 9 by flash chromatography^[13]. Instead, the mixture of 8 and 9 was resolved kinetically by the addition of another alkyne in substoichiometric amounts (0.3-0.4 equiv.). This alkyne reacted preferentially with the minor compound 9 by converting it into the biscoupling product 9'. Much less coupling occurred with the major compound 8 leading to the formation of 8'. Consequently, the desired monocoupling products 8 emerged essentially unaltered from this operation and were now easily separable. Their yields ranged from 36 to 52% based on the starting bistriflate 3; they would be 52-74% if referred to the initially used alkyne and if complete incorporation of the second alkyne into the scavenging products 8' or 9' was assumed.

Table 1. Selected ¹H- and ¹³C-NMR data of the monocoupling products 8

	δ(¹ H)			⁵ J _{1',4'}	δ	(¹³ C)
8	2-Н	1'-H	4'-H ₂	[Hz]	C-2	C-1'
а	6.19	5.49	4.38;4.46	2.3;2.2	129.15	97.04 or 98.98
b	6.15	5.41	4.55	2.1	129.07	98.92
c	6.16	5.46	2.64	2.7	128.57	99.58
d	6.12	5.42	2.47	2.5	128.39	99.80
e	6.07	5.45	1.67	2.9	127.16	100.90
f	6.23	5.49	-	-	128.63	99.53

The semicyclic C = C bond of all monocoupling products 8 must be (Z)-configurated. This follows from their role as (major) intermediates on the way to the previously described biscoupling products 7 which displayed (Z)-configuration according to NOE experiments [5]. The positional selectivity of the coupling reaction – attachment of the alkyne to the semicyclic sp²-carbon atom of 8 – is inferred from the ¹H-NMR spectra (Table 1). In 8a-e, the propargyl protons experience a long-range coupling J = 2.2-2.9 Hz across the $C \equiv C$ bond. It is caused by the olefinic 1'-H as explicitly shown for 8a or b by selective decoupling. If 8 carried the alkynyl chain at the site of the former endocyclic triflate, no such coupling would be observed. The monocoupling product 8f is devoid of propargyl protons. Therefore, its regiochemistry could not be determined by the long-range coupling criterion. Nonetheless, it should be analogous to the regiochemistry encountered before because of the similarity of all ¹H- and ¹³C-NMR data in the series (selected values: Table 1).

With monocoupling products 8 at hand, the stage was set for coupling with the second alkyne. This reaction again proceeded under Cacchi-like Pd(0) catalysis conditions. It afforded, with complete control of regionselectivity and retention of (Z) stereochemistry (vide infra), the biscoupling

2	R ¹	R ²	Yield
а	ОТНР	SiMe ₃	77%
b	ОТНР	CH ₂ SiMe ₃	52%
c	OTHP	nC ₄ H ₉	29%
d	OSiPh ₂ Me	CH ₂ SiMe ₃	55%
e	CH₂OH	SiMe ₃	73%
f	CH₂OH	CH ₂ SiMe ₃	65%
g	CH₂CH₂OH	SiMe ₃	72%
h	SiMe ₃	CH₂CH₂OH	60%
i	a)	CH ₂ CH ₂ OH	60%
j	a)	CH2CH2CH2OH	57%

^{a)} CH_2 - R^1 = $SiMe_3$.

products 2 containing differentiated alkynyl groups. Yields were 29-77% and might be higher if we were able to purify these exceedingly labile compounds without losses caused by decomposition.

We assigned the (Z) configuration to the biscoupling products 2 since the similarly prepared biscoupling products 7 were *proven* to possess (Z) geometry^[5]. In addition, direct support for this assignment was obtained for 2b. Thus, irradiation of the semicyclic sp²-H $(\delta = 5.41)$ enhanced absorption of 4-H₂ $(\delta = 2.56-2.68)$ by 4% due to a nuclear Overhauser effect. This proves the proximity of these entities and consequently the (Z) geometry.

Table 2. Selected ¹H- and ¹³C-NMR data of the biscoupling products 2

		δ(¹ H)		⁵ J _{1',4'}	δ(¹³ (C)
2	2-H	1'-H	4'-H ₂	[Hz]	C-2	C-1'
а	6.65	5.48	4.35;4.42	2.5;2.5	151.78	98.47
b	6.38	5.41	4.37;4.44	2.4;2.3	147.54	97.23
c	6.49	5.45	4.38;4.45	2.5;2.3	148.35	97.71
d	6.40	5.38	4.54	2.3	146.95	97.62
e	6.66	5.44	superimposed	inaccessible	152.08	98.77
f	6.42	5.38	superimposed	inaccessible	147.98	97.86
g	6.63	5.42	superimposed	inaccessible	151.53	99.10
h	6.46	5.45	1.68	2.9	147.92	99.66
i	6.55	5.50	-	-	149.92	98.47
j	6.51	5.48	-	-	149.38	98.35
1 ^[2a]	6.79	5.82	4.14	not resolved	139.4	106.5

The ¹H- and ¹³C-NMR spectra of 2a-j are consistent with the suggested structures (for representative data see Table 2). In compounds 2a-d and h, the location of the different alkyne residues are revealed by couplings across one of the C \equiv C bonds (${}^5J_{H,H} = 2.3 - 2.9$ Hz). These couplings invariably concern the alkyne introduced first into the bistriflate 3. This underlines the earlier observation that coupling occurs preferentially at the least hindered position. The regiochemistry of biscoupling products 2e-g could not be determined similarly since here the propargyl signals were obscured by other resonances, and no splitting due to ${}^5J_{\rm H,H}$ could be observed. Also, 2i and j do not lend themselves to this kind of structural proof due to the absence of propargyl protons in the first reactant, trimethylsilylacetylene. Nonetheless, these compounds' precursors of established structures 8c,d and f leave no doubt as to the suggested regiochemistry of the derived biscoupling products. The same applies to the consistency of all ¹H- and ¹³C-NMR shifts within the series (Table 2).

In summary, dienediynes 2 with two different alkyne substituents have been prepared stereoselectively from formyl-cyclopentanone in 4 steps. The only previous synthesis of a compound of type 2 with (Z) configuration was achieved by Terashima et al. ^[4g]; it involved 9 steps starting from oxocyclopentanecarboxylic ester and the introduction of the stereogenic double bond with 37:63 selectivity.

We are indebted to Prof. S. Berger for his help in recording and assigning the NMR spectra, and to Dr. G. Lange and F. Dadrich for measuring the high-resolution mass spectra. To Dr. D. Scheutzow, E. Ruckdeschel, and A. Mbonimana, we express our gratitude for additional NMR- and to F. Schmock and S. Schneider for IR-spectroscopic assistance. Financial support from the Fonds der Chemischen Industrie and donations of triflic anhydride from Merck-Schuchardt, PdCl₂ from Degussa, and n- and tBuLi from Chemetall GmbH are gratefully acknowledged. S. W. S. thanks the Graduiertenkolleg Marburg for a fellowship.

Experimental

¹H NMR (tetramethylsilane as internal standard in CDCl₃) and ¹³C NMR (CDCl₃ as internal standard and solvent): Bruker AC 300, AC 200, WM 400, or WH 400; coupling constants in Hz. -MS: MAT90, 8200, FINNIGAN MAT. - IR (CDCl₃ solution): Bruker IFF88. — All reactions were performed in oven-dried (100 °C) glassware under dry nitrogen. Coupling reactions were carried out in Schlenk tubes. - THF was freshly distilled from K/Na, HN(iPr)₂ from CaH₂. - All products were purified by flash chromatography^[13] on silica gel 60 (Merck; particle size 0.040-0.063 mm, 230-400 mesh ASTM); for each separation the column diameter, the fraction size, the eluent, and the productcontaining fractions are given in parentheses [i.e.: 2 cm, 15 ml, diethyl ether/petroleum ether (1:10), F 2-3]. — Unless stated otherwise, the new compounds are colorless to yellow oils solidifying only in the freezer (< -30°C) if at all. They are highly labile and start to decompose at room temperature already within minutes (unless kept in solution). Therefore, the removal of the eluent from 2 and 8 had to be performed at 0°C/0.03 Torr and could not always be brought to completion prior to loss of product through decomposition. Accordingly, the yields of 2 and 8 are corrected for up to 21 mol-% of residual eluent. As another consequence, correct combustion analyses of 2 and 8 could not be obtained.

Alkynes: Trimethylsilylacetylene, Trimethylpropargylsilane, 3-Butyn-1-ol, 4-Pentyn-1-ol, 1-Hexyne, 2-Methyl-3-butyn-2-ol, and Propargyl Alcohol are commercially available.

2-(Propargyloxy) tetrahydropyran was prepared following the general procedure of ref.^[14].

(Z)-5-[(Trifluoromethanesulfonyloxy)methylene]-1-cyclopentenyl Trifluoromethanesulfonate (3) via (Z)-(2-Oxocyclopentylidene) methyl Trifluoromethanesulfonate (6): At -65°C tBuLi (8.75 ml of a 1.28 M solution in hexanes, 11.2 mmol, 1.0 equiv.) was added very slowly to 2-formylcyclopentanone (4; 1.27 g, 11.2 mmol)^[4h,8] in THF (70 ml) with vigorous stirring. After 15 min, (CF₃SO₂)₂O (1.84 ml, 11.2 mmol, 1.0 equiv.) was added. After a few minutes, the cold reaction mixture was poured into satd. aq. NaHCO₃/brine (1:2). Extraction with diethyl ether and flash chromatography [3.5] cm, 60 ml, tBuOMe/petroleum ether (1:5 \rightarrow 1:1), F 5-12] gave 6 (1.16 g, ca. 4.73 mmol) as a brown oil which, because of its instability, was immediately dissolved in THF (10 ml) and added dropwise at -65°C to a stirred solution of LiN(SiMe₃)₂ [prepared at -65°C from a THF solution (40 ml) of HN(SiMe₃)₂ (1.18 ml, 5.66 mmol, ≥ 1.2 equiv.) and nBuLi (2.36 ml of a 2.20 M solution in hexanes, 5.19 mmol, ≥ 1.1 equiv.)]. After 5 min, (CF₃SO₂)₂O (0.78 ml, 4.74 mmol, ≥1.0 equiv.) was added. The resulting mixture was poured cold into satd. aq. NaHCO3/brine (1:2) and this mixture extracted with diethyl ether. Flash chromatography [3.5 cm, 60 ml, tBuOMe/petroleum ether (1:10), F 6-14] gave 3 (1.20 g, 29% based on 1] as a brown oil. Repeated recrystallization of this oil from pentane at -35° C yielded colorless crystals (m.p. ca. -5-0°C) for the X-ray analysis.

3: 1 H NMR (400 MHz): $\delta = 2.61 - 2.65$ (m, 3-H₂), 2.73 - 2.76 (m, 4-H₂), 6.23 (m_e, 2-H), 6.54 (m_e, 1'-H). — NOE experiments: Irradiation at $\delta = 6.54$ resulted in no (!) signal enhancement, whereas irradiation at $\delta = 6.23$ enhanced the absorption at $\delta = 2.61 - 2.65$ by 2.5%. — H,H-decoupling experiments: Irradiation at $\delta = 6.54$ converted the m_c at $\delta = 6.23$ into a triplet (J = 3.0), and irradiation at $\delta = 6.23$ caused a narrowing of the m_c at $\delta = 6.54$. — 13 C NMR (gated decoupling, 101 MHz): $\delta = 24.43$ (t with further unresolved splitting, $^{1}J_{\rm C,H} = 134.4$, C-3 or C-4), 26.56 (t with further unresolved splitting, $^{1}J_{\rm C,H} = 136.1$, C-3 or C-4), 118.46 (q, $^{1}J_{\rm C,F} = 320.6$, CF₃), 118.52 (q, $^{1}J_{\rm C,F} = 321.0$, CF₃), 126.24 (d, $^{1}J_{\rm C,H} = 205.0$, C-1'), 126.98 (m_e, C-5), 129.99 (dt, $^{1}J_{\rm C,H} = 172.5$, $^{2}J_{\rm C,H} = 6.4$, C-2), 144.98 (m_e, C-1). — Heterodecoupling experiment: Irradiation at δ (¹H) = 6.23 decoupled and enhanced the signal at δ (¹³C) = 129.99. — IR: $\tilde{\nu} = 3110$ cm⁻¹, 2920, 1675, 1620, 1430, 1220, 1140, 1080, 1020.

C₈H₆F₆O₆S₂ Calcd. 375.9510 Found 375.9517 (EI HRMS)

6: ¹H NMR (300 MHz): $\delta = 2.03$ (tt, $J_{4',3'} = J_{4',5'} = 7.4$, 4'-H₂), 2.41 (t, $J_{3',4'} = 7.8$, 3'-H₂), 2.71 (td, $J_{5',4'} = 7.2$, $J_{5',1} = 2.3$, 5'-H₂), 6.62 (t, $J_{1,5'} = 2.4$, 1-H). - ¹³C NMR (50 MHz): $\delta = 20.35$, 27.05, 39.44 (C-3', C-4', C-5'), 118.41 (q, ${}^{1}J_{\text{C,F}} = 321$, CF₃), 126.27 (C-1'), 133.52 (C-1), 200.66 (C-2'). - IR: $\tilde{v} = 2950$ cm⁻¹, 1735, 1650, 1430, 1225, 1215, 1140, 1030.

Monocoupling Products 8

(Z)-5-[4-(2-Tetrahydropyranyloxy)-2-butynylidene]-1-cyclopentenyl Trifluoromethanesulfonate (8a): Pd(PPh₃)₂Cl₂ (79.4 mg, 0.113 mmol, 0.05 equiv.), CuI (64.7 mg, 0.340 mmol, 0.15 equiv.), and HN(iPr)₂ (3.5 ml) were successively added at room temperature to 3 (852 mg, 2.26 mmol) and THP-protected propargyl alcohol (349 mg, 2.49 mmol, 1.1 equiv.) in THF (14 ml). After stirring for 55 min, trimethylsilylacetylene (96.0 μ l, 0.679 mmol, 0.3 equiv.) was added. After 2 h, the mixture was poured into satd. NaCl. The product was extracted with diethyl ether. Drying of the extract with Na₂SO₄ and charcoal, filtration through a pad of Celite, and flash

Stereoselective 275

chromatography [2 cm, 20 ml, pentane → diethyl ether/pentane (1:3), F 18-27] furnished **8a** (409 mg, 47%), - ¹H NMR (200 MHz): $\delta = 1.40 - 1.97$ (m, 3"-H₂, 4"-H₂, 5"-H₂), 2.47 - 2.60 and 2.67 - 2.81 (2 m, $3-H_2$, $4-H_2$), 3.45 - 3.60 and 3.76 - 3.94 (2 m, 6"- H_2), AB signal ($\delta_A = 4.38$, $\delta_B = 4.46$, $J_{AB} = 15.9$, in addition split by $J_{A,1'} = 2.3$, $J_{B,1'} = 2.2$, 4'-H₂), 4.81 (dd, $J_{2^{\circ},3^{\circ}(A)} = J_{2^{\circ},3^{\circ}(B)} = 3.2$, 2"-H), 5.49 (m_c, 1'-H), 6.19 (m_c, 2-H). — Homodecoupling experiments for the assignments of 2-H and 1'-H: Splitting by $J_{4',1'}$ at $\delta(4'$ - H_2) vanished during irradiation at $\delta = 5.49$ but remained unchanged during irradiation at $\delta = 6.19$. - ¹³C NMR (50 MHz): δ = 19.12, 22.45, 25.42, 26.24, 28.73, 30.31 (C-3, C-4, C-3", C-4", C-4", C-4") 5"), 55.11, 61.96 (C-4', C-6"), 80.18, 91.76 (sp C), 97.04, 98.98 (C-1', C-2"), 118.50 (q, ${}^{1}J_{CF} = 318.9$, CF₃), 129.15 (C-2), 146.50, 148.99 (C-1, C-5); impurity signal at $\delta = 124.85$. – IR: $\tilde{v} = 2930$ cm⁻¹. 1745, 1430, 1240, 1215, 1185, 1140, 1120, 1070, 1055, 1040, 1025, 870, 855, 770, 600.

C₁₅H₁₇F₃O₅S Calcd. 366.0749 Found 366.0762 (EI HRMS)

(Z)-5-[4-(Methyldiphenylsilyloxy)-2-butynylidene]-1-cyclopentenyl Trifluoromethanesulfonate (8b): Pd(PPh₃)₂Cl₂ (84.0 mg, 0.120 mmol, 0.1 equiv.) and CuI (68.3 mg, 0.359 mmol, 0.3 equiv.) were added at room temperature to 3 (450 mg, 1.20 mmol) and propargyl alcohol (0.21 ml, 3.60 mmol, 3.0 equiv.) in THF (5 ml) followed by the addition of HN(iPr)₂ (1.5 ml). After 50 min, the reaction mixture was worked up as described for the synthesis of 8a. Flash chromatography [2 cm, 20 ml, diethyl ether/pentane (1:1), F 6-11] yielded a severely contaminated monocoupling product (207 mg). It was dissolved in CH₂Cl₂ (4 ml) and the solution added at room temperature to imidazole (149 mg, 2.20 mmol, ≥3.1 equiv.) and chloro(methyl)diphenylsilane (0.23 ml, 1.10 mmol, ≥1.5 equiv.) in CH₂Cl₂ (4 ml). After 3.5 h, more imidazole (99.6 mg, 1.46 mmol, ≥2.0 equiv.) and chloro(methyl)diphenylsilane (0.15 ml, 0.73 mmol, ≥1.0 equiv.) were added. After 2 h, the mixture was extracted with diethyl ether/H₂O and purified by flash chromatography [2 cm, 20 ml, pentane \rightarrow diethyl ether/pentane (1:10), F 20-24] to give **8b** (206 mg, 35% from 3). - ¹H NMR (200 MHz): $\delta = 0.71$ [s, $Si(CH_3)Ph_2$, 2.43 – 2.55 and 2.63 – 2.76 (2 m, 3-H₂, 4-H₂), 4.55 (d, $J_{4',1'} = 2.1, 4'-H_2$, 5.41 (m_c, 1'-H), 6.15 (m_c, 2-H), 7.29 – 7.46 and 7.56-7.69 (2 m, 6 and 4H, respectively; 2 Ph). - Homodecoupling experiments for the assignments of 2-H and 1'-H: Splitting by $J_{4',1'}$ at $\delta(4'-H_2)$ vanished during irradiation at $\delta = 5.41$ but remained unchanged during irradiation at $\delta = 6.15$. - ¹³C NMR (50 MHz): $\delta = -2.99 \text{ (Si}CH_3Ph_2), 26.11, 28.61 (C-3, C-4), 52.65 (C-4'), 80.86,$ 93.71 (sp C), 98.92 (C-1'), 118.53 (q, ${}^{1}J_{CF} = 321.1$, CF₃), 127.76, 129.79, 134.39, 135.49 (Ph), 129.07 (half as intense as 129.79, C-2), 146.23, 149.01 (C-1, C-5). – MS [CI (NH₃)]: m/z (%) = 496 (89) $[M^{+} + NH_{4}^{+}]$, 479 (70) $[M^{+} + H^{+}]$; HRMS by CI was impossible for technical reasons and also by EI due to rapid decomposition of the sample at the required temperature.

(Z)-5-(5-Hydroxy-2-pentynylidene)-1-cyclopentenyl Trifluoromethanesulfonate (8c) was prepared from 3 (400 mg, 1.06 mmol), Pd(PPh₃)₂Cl₂ (37.3 mg, 0.053 mmol, 0.05 equiv.), CuI (30.3 mg, 0.159 mmol, 0.15 equiv.), and 3-butyn-1-ol (88.4 µl, 1.17 mmol, 1.10 equiv.; coupling time 4 h) as described for the preparation of 8a except for the use of 0.40 equiv. rather than 0.30 equiv. of trimethylsilylacetylene (60.0 µl, 0.425 mmol; coupling time 4.5 h). Flash chromatography [2 cm, 20 ml, diethyl ether/pentane (1:1), F 10-16] gave the product (150 mg, 36%). — ¹H NMR (200 MHz): δ = 1.83 (t, $J_{OH,5}$ = 6.5, OH), 2.48-2.59 and 2.69-2.79 (2 m, 3-H₂, 4-H₂), 2.64 (td, $J_{4',5'}$ = 6.1, $J_{4',1'}$ = 2.6, 4'-H₂), 3.75 (dt, $J_{5',OH}$ = $J_{5',4'}$ = 6.2, 5'-H₂), 5.46 (m_c, 1'-H), 6.16 (m_c, 2-H). — ¹³C NMR (50 MHz): δ = 24.30, 26.14, 28.69 (C-3, C-4, C-4'), 60.93 (C-5'), 78.00, 93.41 (sp C), 99.58 (C-1'), 118.72 (q, ${}^{1}J_{C,F}$ = 327.6, CF₃), 128.57 (C-2), 145.41,

149.32 (C-1, C-5). - IR: $\tilde{v} = 3620 \text{ cm}^{-1}$, 3020, 2930, 2340, 1605, 1425, 1240, 1220, 1180, 1140, 1070, 1050, 870, 855, 775, 620, 600.

C₁₁H₁₁F₃O₄S Calcd. 296.0330 Found 296.0319 (EI HRMS)

(Z)-5-(6-Hydroxy-2-hexynylidene)-1-cyclopentenyl Trifluoromethanesulfonate (8d) was prepared from 3 (229 mg, 0.609 mmol). Pd(PPh₃)₂Cl₂ (21.4 mg, 0.030 mmol, 0.05 equiv.), CuI (17.4 mg, 0.091 mmol, 0.15 equiv.), 4-pentyn-1-ol (62.0 µl, 0.670 mmol, 1.1 equiv.; coupling time 4 h), and trimethylsilylacetylene (34.4 µl, 0.244 mmol, 0.4 equiv.; coupling time 2 h) as described for the preparation of 8a. Flash chromatography [2 cm, 20 ml, diethyl ether/pentane (1:1 \rightarrow 2:1), F 9-11] furnished the product (87.2 mg, 37%). - ¹H NMR (200 MHz): $\delta = 1.77$ [tt, $J_{5',6'} = J_{5',4'} = 6.5$, 5'-H₂, superimposing signal at $\delta = 1.80$ (s, OH)], 2.41-2.56 and 2.66-2.77[2 m, 3-H₂, 4-H₂, superimposed in part by signal at $\delta = 2.47$ (td, $J_{4'.5'} \approx 7.0, J_{4'.1'} \approx 2.5, 4'-H_2$, 3.73 (t, $J_{6'.5'} = 6.1, 6'-H_2$), 5.42 (m_c, 1'-H), 6.12 (m_c, 2-H). - ¹³C NMR (50 MHz): δ = 16.27, 26.09, 28.61, 31.09 (C-3, C-4, C-4', C-5'), 61.69 (C-6'), 76.76, 96.35 (sp C), 99.80 (C-1'), 128.39 (C-2), 144.95 (C-5 and C-1 because of extra intensity); the signals of the CF₃ group did not emerge from the spectral noise. – IR: $\tilde{v} = 3625 \text{ cm}^{-1}$, 2930, 2340, 2250, 1605, 1425, 1265, 1240, 1220, 1180, 1140, 1070, 1005, 870, 855, 820, 620, 600.

C₁₂H₁₃F₃O₄S Calcd. 310.0487 Found 310.0483 (EI HRMS)

(Z)-5-[4-(Trimethylsilyl)-2-butynylidene]-1-cyclopentenyl Trifluoromethanesulfonate (8e) was prepared from 3 (121 mg, 0.321 mmol), Pd(PPh₃)₂Cl₂ (11.3 mg, 0.016 mmol, 0.05 equiv.), CuI (9.2 mg, 0.048 mmol, 0.15 equiv.), and propargyltrimethylsilane (52.6 µl, 0.353 mmol, 1.1 equiv.; coupling time 3.5 h) according to the procedure described for the preparation of 8a with the following exception: The addition of trimethylsilylacetylene was replaced by the addition of 2-methyl-3-butyn-2-ol (10.0 µl, 0.096 mmol, 0.3 equiv.; coupling time 2.5 h). Flash chromatography [2 cm, 20 ml, pentane → diethyl ether/pentane (1:20), F 12-13] yielded the product [46.6 mg, slightly contaminated with unknown compound(s), 37%]. - ¹H NMR (200 MHz): $\delta = 0.10$ [s, Si(CH₃)₃], 1.67 (d, $J_{4',1'} = 2.9$, $4'-H_2$), 2.44 – 2.56 and 2.65 – 2.76 (2 m, 3-H₂, 4-H₂), 5.45 (m_c, 1'-H), 6.07 (m_c, 2-H). - ¹³C NMR (50 MHz): δ = -2.00 [Si(CH₃)₃], 8.56 (C-4'), 26.05, 28.57 (C-3, C-4), 75.36, 96.04 (sp C), 100.90 (C-1'), 118.78 (q, ${}^{1}J_{CF} = 323.6$, CF₃), 127.16 (C-2), 142.93, 149.60 (C-1, C-5). – IR: $\tilde{v} = 2960 \text{ cm}^{-1}$, 2930, 2855, 2245, 2200, 2150, 1605, 1425, 1250, 1215, 1170, 1140, 1090, 1070, 850, 770, 620, 600.

$C_{13}H_{17}F_3O_3SSi$ Calcd. 338.0620 Found 338.0619 (EI HRMS)

(Z)-5-[3-(Trimethylsilyl)-2-propynylidene]-1-cyclopentenyl Trifluoromethanesulfonate (8f) was prepared from 3 (95.1 mg, 0.253 mmol), Pd(PPh₃)₂Cl₂ (8.9 mg, 0.013 mmol, 0.05 equiv.), CuI (7.2 mg, 0.038 mmol, 0.15 equiv.), trimethylsilylacetylene (39.3 µl, 0.278 mmol, 1.1 equiv.; coupling time 1.5 h), and 2-methyl-3-butyn-2-ol (7.3 µl, 0.076 mmol, 0.3 equiv.; coupling time 3 h) as described for the preparation of 8e. Flash chromatography [2 cm, 20 ml, pentane \rightarrow diethyl ether/pentane (1:10), F 10-11] yielded the product (42.4 mg, 52%). - ¹H NMR (200 MHz): $\delta = 0.18 \text{ [s, Si(CH₃)₃]}$, 2.48 - 2.59 and 2.67 - 2.79 (2 m, $3-H_2$, $4-H_2$), 5.49 (m_c, 1'-H), 6.23 $(m_{c_1} 2-H)$. - ¹³C NMR (50 MHz): $\delta = -0.32$ [Si(CH₃)₃], 26.30, 28.82 (C-3, C-4), 99.53 (C-1'), 100.19 (1 sp C), 118.68 (q, ${}^{1}J_{CF}$ = 322.0, CF₃), 128.63 (C-2), 146.86, 149.22 (C-1, C-5); the signal of the second sp C was obscured by the spectral noise. – IR: \tilde{v} = 2960 cm⁻¹, 2925, 1425, 1250, 1240, 1215, 1175, 1140, 1090, 1060, 1035, 850, 765, 600.

C₁₂H₁₅F₃O₃SSi Calcd. 324.0463 Found 324.0463 (EI HRMS) Biscoupling Products 2^[15]

(Z)-5-[4-(2-Tetrahydropyranyloxy)-2-butynylidene]-1-[2-(tri-methylsilyl)-1-ethynyl]-1-cyclopentene (2a): Pd(PPh₃)₂Cl₂ (36.2 mg,

0.052 mmol, 0.2 equiv.), CuI (14.7 mg, 0.077 mmol, 0.3 equiv.), and HN(iPr)₂ (1 ml) were successively added at room temperature to 8a (94.5 mg, 0.258 mmol) and trimethylsilylacetylene (72.9 μl, 0.516 mmol, 2.0 equiv.) in THF (3 ml). After 5.5 h, the reaction mixture was extracted with brine/diethyl ether. Flash chromatography [2 cm, 20 ml, diethyl ether/pentane (1:10 \rightarrow 1:5), F 5-7] yielded the product (62.6 mg, 77%). - ¹H NMR (200 MHz): $\delta = 0.20$ [s, $Si(CH_3)_3$, 1.40-1.95 (m, 3"-H₂, 4"-H₂, 5"-H₂), 2.42-2.54 and 2.57 - 2.70 (2 m, $3-H_2$, $4-H_2$), 3.43 - 3.58 and 3.76 - 3.92 (2 m, 6"- H_2), AB signal ($\delta_A = 4.35$, $\delta_B = 4.42$, $J_{A,B} = 15.6$, in addition split by $J_{A,1'}=J_{B,1'}=2.5$, 4'-H₂), 4.81 (dd, $J_{2^{\circ},3^{\circ}(A)}\approx J_{2^{\circ},3^{\circ}(B)}\approx 3.2$, 2"-H), 5.48 (m_c, 1'-H), 6.65 (m_c, 2-H). - ¹³C NMR (50 MHz): $\delta=$ -0.09 [Si(CH₃)₃], 19.14, 25.39, 30.30, 30.69, 31.45 (C-3, C-4, C-3", C-4", C-5"), 55.23 (C-4'), 61.93 (C-6"), 82.76, 89.07, 98.77, 99.14 (sp C), 96.83 (C-2"*), 98.47 (C-1'*), 126.73 (C-1), 151.78 (C-2), 154.09 (C-5): * assigned in analogy to 2b. – IR: $\tilde{v} = 2945 \text{ cm}^{-1}$, 2875, 2855, 2245, 2150, 1435, 1345, 1250, 1200, 1120, 1075, 1055, 1040, 1025, 840, 815, 520.

C₁₉H₂₆O₂Si Calcd. 314.1702 Found 314.1698 (EI HRMS)

(Z)-5-[4-(2-Tetrahydropyranyloxy)-2-butynylidene]-1-[3-(trimethylsilyl)-1-propynyl]-1-cyclopentene (2b) was prepared from 8a (222 mg, 0.605 mmol), propargyltrimethylsilane (0.18 ml, 1.20 mmol, 2.0 equiv.), Pd(PPh₃)₂Cl₂ (42.5 mg, 0.061 mmol, 0.1 equiv.), and CuI (34.6 mg, 0.182 mmol, 0.3 equiv.) as described for the preparation of 2a. The total reaction time was 7 h, but after 3 h more Pd(PPh₃)₂Cl₂ (21.2 mg, 0.030 mmol, 0.05 equiv.) was added. Flash chromatography [2 cm, 20 ml, diethyl ether/pentane (1:10 \rightarrow 1:4), F 8-10] yielded the product (103 mg, 52%). - ¹H NMR (200 MHz): $\delta = 0.12$ [s, Si(CH₃)₃], 1.44-1.90 [m, 3"-H₂, 4"-H₂, 5"-H₂, superimposing signal at $\delta = 1.69$ (s, 3"-H₂)], 2.39 – 2.51 (m, 3-H₂), 2.56-2.68 (m, $4-H_2$), 3.43-3.58 and 3.75-3.91 (2 m, $6''-H_2$), AB signal ($\delta_A = 4.37$, $\delta_B = 4.44$, $J_{A,B} = 15.6$, in addition split by $J_{A,1} = 4.44$ 2.4, $J_{\text{B,1'}} = 2.3$, 4'-H₂), 4.84 (dd, $J_{2^{\circ},3^{\circ}(\text{A})} \approx J_{2^{\circ},3^{\circ}(\text{B})} \approx 3.1$, 2"-H), 5.41 (m_c, 1'-H), 6.38 (m_c, 2-H). - Homodecoupling experiments for the assignments of 2-H and 1'-H: Splitting by $J_{4',1'}$ at $\delta(4'-H_2)$ vanished during irradiation at $\delta = 5.41$ but remained unchanged during irradiation at $\delta = 6.38$. NOE experiments for the assignment of 3-H₂ vs. 4-H₂ and for assignment of the (Z) configuration: Irradiation at $\delta = 5.41$ enhanced absorption at $\delta = 2.56 - 2.68$ by 4%; irradiation at $\delta = 6.38$ enhanced absorption at $\delta = 2.39 - 2.51$ by the same amount. - ¹³C NMR (101 MHz, gated decoupling): δ = -1.99 [q, ${}^{1}J_{C,H} = 119.3$, Si(CH₃)₃], 7.96 (t, ${}^{1}J_{C,H} = 125.1$, 3"'-C), 18.98 (t, ${}^{1}J_{C,H} = 128.5$) and 25.24 (t, ${}^{1}J_{C,H} = 125.9$) and 28.5 – 32.5 (m, C-3, C-4, C-3", C-4", C-5"), 55.10 (t with further unresolved splittings, ${}^{1}J_{C,H} = 148.0, C-4'$), 61.85 (t with further unresolved splittings, ${}^{1}J_{C,H} = 143.1$, C-6"), 73.39, 82.98 (2 s, C-2', C-1""), 89.23 [m_c (perhaps unresolved dt), C-3'], 93.20 (t, ${}^{2}J_{\text{C-2''},3''-\text{H}} = 11.1$, C-2'''), 96.48 (d, ${}^{1}J_{C,H} = 165.6$, C-2"), 97.23 (d, ${}^{1}J_{C,H} = 161.7$, C-1'), 127.25 (s, C-1), 147.54 (d, ${}^{1}J_{C,H} = 166.3$, C-2), 156.15 (s, C-5). — Heterodecoupling experiments for the assignment of individual $\delta(^{13}C)$: Irradiation at $\delta(^{1}H) = 4.37/4.44$ decoupled and enhanced signal at $\delta(^{13}C) = 55.10$ and simplified signal at $\delta(^{13}C) = 89.23$ to a very closely spaced doublet (tentatively assigned); irradiation at $\delta(^{1}H)$ = 5.41 decoupled and enhanced signal at $\delta(^{13}C) = 97.23$; irradiation at $\delta(^{1}H) = 6.38$ decoupled and enhanced signal at $\delta(^{13}C) =$ $147.54. - IR: \tilde{v} = 2945 \text{ cm}^{-1}, 2875, 2855, 2245, 2205, 1345, 1250,$ 1200, 1120, 1075, 1055, 1025, 845, 815.

C₂₀H₂₈O₂Si Calcd. 328.1858 Found 328.1861 (EI HRMS)

(Z)-1-(1-Hexynyl)-5-[4-(2-tetrahydropyranyloxy)-2-butynylidene]-1-cyclopentene (2c) was prepared from 8a (171 mg, 0.466 mmol), 1-hexyne (80.2 μ l, 0.699 mmol, 1.5 equiv.), Pd(PPh₃)₂Cl₂ (32.7 mg, 0.047 mmol, 0.1 equiv.), and CuI (26.6 mg, 0.140 mmol,

0.3 equiv.) as described for the preparation of 2a. The total reaction time was 73 h. Flash chromatography [2 cm, 20 ml, diethyl ether/ pentane (1:10 \rightarrow 1:5), F 11-14] furnished the product (40.7 mg, 29%). – ¹H NMR (200 MHz): $\delta = 0.92$ (t, $J_{6'',5''} = 7.1$, $6'''-H_3$), 1.33-1.90 (m, 3"-H₂, 4"-H₂, 5"-H₂, 4"'-H₂, 5"'-H₂), 2.39 (t, $J_{3",4"}$ = 7.0, $3'''-H_2$), 2.43-2.53 and 2.60-2.70 (2 m, $3-H_2$, $4-H_2$), 3.45-3.59and 3.78 - 3.92 (2 m, 6"-H₂), AB signal ($\delta_A = 4.38$, $\delta_B = 4.45$, $J_{A,B} =$ 15.6, in addition split by $J_{A,1'} = 2.5$, $J_{B,1'} = 2.3$, 4'-H₂), 4.85 (dd, $J_{2'',3''(A)} \approx J_{2'',3''(B)} \approx 3.2, 2''-H), 5.45 \text{ (m}_c, 1'-H), 6.49 \text{ (m}_c, 2-H); im$ purity signal at $\delta = 1.25$ (s). - ¹³C NMR (50 MHz): $\delta = 13.63$ (C-6"), 19.15, 19.33, 22.15, 25.39, 30.34, 30.49, 30.76, 31.28 (C-3, C-4, C-3", C-4", C-5", C-3"', C-4"', C-5"'), 55.17 (C-4'), 61.93 (C-6"), 74.57, 83.00, 89.58, 95.10 (sp C), 96.68 (C-2"*), 97.71 (C-1'*), 127.10 (C-1), 148.35 (C-2), 155.59 (C-5); * assigned in analogy to 2b. – IR: \tilde{v} = 2935 cm⁻¹, 2875, 2245, 1710, 1610, 1445, 1355, 1345, 1265, 1200, 1185, 1120, 1075, 1055, 1035, 1020, 815.

C₂₀H₂₆O₂ Calcd. 298.1933 Found 298.1927 (EI HRMS)

(Z)-5-[4-(Methyldiphenylsilyloxy)-2-butynylidene]-1-[3-(trimethylsilyl)-1-propynyl]-1-cyclopentene (2d) was prepared from 8b (181 mg, 0.376 mmol), 1-trimethylsilyl-2-propyne (0.11 ml, 0.75 mmol, 2.0 equiv.), Pd(PPh₃)₂Cl₂ (52.8 mg, 0.075 mmol, 0.2 equiv.), and CuI (21.5 mg, 0.113 mmol, 0.3 equiv.) as described for the preparation of 2a. The total reaction time was 5.5 h, but already after 70 min Pd(PPh₃)₂Cl₂ (26.4 mg, 0.038 mmol, 0.1 equiv.) was added. Flash chromatography [2 cm, 20 ml, pentane → diethyl ether/pentane (1:10), F 12-14] yielded the product [95.6 mg, slightly contaminated with unknown compound(s), 55%]. - 1H NMR (200 MHz): $\delta = 0.07$ [s, Si(CH₃)Ph₂], 0.13 [s, Si(CH₃)₃], 1.57 (s, 3"-H₂), 2.41 - 2.53 and 2.58 - 2.70 (2 m, $3 - H_2$, $4 - H_2$), 4.54 (d, $J_{4',1'} = 2.3$, 4' - H_2), 5.38 (m_c, 1'-H), 6.40 (m_c, 2-H), 7.30 – 7.47 and 7.58 – 7.68 (2 m, 6H and 4H, respectively, 2 Ph). - ¹³C NMR (50 MHz): $\delta = -2.64$ $[Si(CH_3)Ph_2]$, -1.99 $[Si(CH_3)_3]$, 7.89 (C-3"), 30.36, 31.28 (C-3, C-4), 52.89 (C-4'), 64.63, 74.43, 82.85, 91.72 or 93.08 (sp C), 97.62 (C-1'), 125.48 (C-1), 127.82, 129.85, 134.48, 135.56 [Si(CH₃)Ph₂], 146.95 (C-2), 155.53 (C-5). — HRMS: Neither EI (70 eV) nor CI (NH₃) nor CI (isobutane) gave an [M+] peak.

(Z)-5-(5-Hydroxy-2-pentynylidene)-1-<math>[2-(trimethylsilyl)-1ethynyl]-1-cyclopentene (2e) was prepared in 3.5 h from 8c (64.2 mg, 0.216 mmol), trimethylsilylacetylene (0.15 ml, 1.1 mmol, 5.0 equiv.), Pd(PPh₃)₂Cl₂ (30.4 mg, 0.043 mmol, 0.2 equiv.), and CuI (12.4 mg, 0.065 mmol, 0.3 equiv.) according to the procedure described for the preparation of 2a. Flash chromatography [1.5 cm, 4 ml, diethyl ether/pentane (1:1), F 9-13] yielded the product (41.6 mg, 73%). - ¹H NMR (200 MHz): $\delta = 0.21$ [s, Si(CH₃)₃], 1.96 (m_e, OH), 2.43 – 2.70 (m, 3-H₂, 4-H₂, 4'-H₂), 3.73 (br. t, $J_{5'.4'} \approx$ 5.7, 5'-H₂), 5.44 (m_c, 1'-H), 6.66 (m_c, 2-H). - ¹³C NMR (50 MHz): $\delta = 0.07 \text{ [Si(CH₃)₃]}, 24.39 (C-4'), 30.70, 31.43 (C-3, C-4), 61.11$ (C-5'), 79.64, 91.10, 99.77 (3 sp C), 98.77 (C-1'), 126.55 (C-1), 152.08 (C-2), 153.08 (C-5); one sp-C signal was obscured by the spectral noise. Heterodecoupling experiments for the assignment of individual $\delta(^{13}C)$ values: Irradiation at $\delta(^{1}H) = 6.66$ decoupled and enhanced signal at $\delta(^{13}C) = 152.08$; irradiation at $\delta(^{1}H) = 5.44$ decoupled and enhanced signal at $\delta(^{13}C) = 98.77$. – IR: $\tilde{v} = 3625$ cm^{-1} , 3580, 2960, 2845, 2245, 2145, 1445, 1425, 1385, 1250, 1110, 1080, 1050, 1005, 845, 815, 520.

C₁₅H₂₀OSi Calcd. 244.1283 Found 244.1272 (EI HRMS)

(Z)-5-(5-Hydroxy-2-pentynylidene)-1-[3-(trimethylsilyl)-1-propynyl]-1-cyclopentene (2f) was prepared from 8c (84.5 mg, 0.285 mmol), 1-trimethylsilyl-2-propyne (0.21 ml, 1.4 mmol, 5.0 equiv.), Pd(PPh₃)₂Cl₂ (40.0 mg, 0.057 mmol, 0.2 equiv.), and CuI (16.3 mg, 0.086 mmol, 0.3 equiv.) as described for the preparation of 2a. The reaction time was 3.5 h. Flash chromatography [1.5 cm, 4 ml, di-

ethyl ether/pentane (1:2 \rightarrow 1:1), F 14–18] gave 48.6 mg (65%). - ¹H NMR (200 MHz): δ = 0.13 [s, Si(CH₃)₃], 1.68 (s, 3″-H₂), 2.13 (br. s, OH), 2.40–2.52 (m, 3-H₂ or 4-H₂), 2.56–2.69 (m, 3-H₂ or 4-H₂, 4′-H₂), 3.71 (m_c, 5′-H₂), 5.38 (m_c, 1′-H), 6.42 (m_c, 2-H). - ¹³C NMR (50 MHz): δ = -1.93 [Si(CH₃)₃], 8.32 (C-3″), 24.54, 30.37, 31.34 (C-3, C-4, C-4′), 61.14 (C-5′), 74.21, 80.00, 90.55, 92.68 (sp C), 97.86 (C-1′), 127.45 (C-1), 147.98 (C-2), 154.50 (C-5). - IR: \tilde{v} = 3620 cm⁻¹, 3545, 2960, 2930, 2885, 2850, 2245, 2225, 1250, 1150, 1050, 850, 815.

C₁₆H₂₂OSi Calcd. 258.1440 Found 258.1444 (EI HRMS)

(Z)-5-(6-Hydroxy-2-hexynylidene)-1-[2-(trimethylsilyl)-1-ethynyl]-1-cyclopentene (2g) was prepared from 8d (84.4 mg, 0.272 mmol), trimethylsilylacetylene (0.19 ml, 1.4 mmol, 5.0 equiv.), Pd(PPh₃)₂Cl₂ (38.2 mg, 0.054 mmol, 0.2 equiv.), and CuI (15.5 mg, 0.082 mmol, 0.3 equiv.) as described for the preparation of 2a. The reaction time was 4.5 h. Flash chromatography [1.5 cm, 4 ml, diethyl ether/pentane (1:2 \rightarrow 2:1), F 16-20] provided 51.2 mg (72%). – ¹H NMR (200 MHz): $\delta = 0.21$ [s, Si(CH₃)₃], 1.71 (br. s, OH), 1.79 (tt, $J_{5',4'} = J_{5',6'} = 6.5$, 5'-H₂), 2.38 – 2.53 (m, 3-H₂ or 4- H_2 , 4'- H_2), 2.56 – 2.69 (m, 3- H_2 or 4- H_2), 3.75 (t, $J_{6',5'} = 6.1$, 6'- H_2), 5.42 (m_e, 1'-H), 6.63 (m_e, 2-H). - ¹³C NMR (50 MHz): $\delta = 0.07$ [Si(CH₃)₃], 16.63, 30.70 (half as intense as the following resonance), 31.37 (C-3, C-4, C-4', C-5'), 61.96 (C-6'), 78.18, 94.13, 98.41, 99.62 (sp C), 99.10 (C-1'), 126.54 (C-1), 151.53 (C-2), 152.44 (C-5). – IR: $\tilde{v} = 3625 \text{ cm}^{-1}$, 2960, 2845, 2340, 2245, 2150, 1675, 1620, 1445, 1430, 1250, 1080, 1005, 845, 815, 520.

C₁₆H₂₂OSi Calcd. 258.1440 Found 258.1428 (EI HRMS)

(Z)-1-(4-Hydroxy-1-butynyl)-5-[4-(trimethylsilyl)-2-butynylidene]-1-cyclopentene (2h) was prepared from 8e (112 mg, 0.332 mmol), 3-butyn-1-ol (50.3 μl, 0.664 mmol, 2.0 equiv.), Pd(PPh₃)₂Cl₂ (23.3 mg, 0.033 mmol, 0.1 equiv.), and CuI (19.0 mg, 0.100 mmol, 0.3 equiv.) as described for the preparation of 2b. During a total reaction time of 19.5 h, additional portions of Pd(PPh₃)₂Cl₂ (23.3 mg, 0.033 mmol, 0.1 equiv.) were added after 1.5 and 4 h. Flash chromatography [2 cm, 20 ml, diethyl ether/pentane (1:10 → 1:1), F 29-31] yielded 53.3 mg (60%). - ¹H NMR (200 MHz): $\delta =$ 0.12 [s, Si(CH₃)₃], 1.68 (d, $J_{4',1'} = 2.9$, 4'-H₂), 2.07 (t, $J_{OH,4'} = 6.7$, OH), 2.41 - 2.53 and 2.58 - 2.70 [2 m, $3-H_2$, $4-H_2$, superimposed in part by signal at $\delta = 2.66$ (t, $J_{3",4"} = 5.8$, $3"-H_2$)], 3.76 (dt, $J_{4",OH} =$ $J_{4",3"} = 6.2, 4"-H_2$, 5.45 (m_c, 1'-H), 6.46 (m_c, 2-H). - ¹³C NMR (50) MHz): $\delta = -1.93$ [Si(CH₃)₃], 8.69 (C-4'), 24.27, 30.46, 31.15 (C-3, C-4, C-3"), 60.93 (C-4"), 76.82, 90.31, 93.31 (3 sp C), 99.66 (C-1'), 126.51 (C-1), 147.92 (C-2), 151.32 (C-5); one sp-C signal was obscured by the spectral noise. – IR: $\tilde{v} = 3430 \text{ cm}^{-1}$, 2960, 2930, 2245, 2200, 1705, 1405, 1250, 1215, 1110, 1050, 850, 770.

C₁₆H₂₂OSi Calcd. 258.1440 Found 258.1430 (EI HRMS)

(Z)-1-(4-Hydroxy-1-butynyl)-5-[3-(trimethylsilyl)-2-propynylidene]-1-cyclopentene (2i) was prepared from 8f (94.7 mg, 0.292 mmol), 3-butyn-1-ol (44.2 μl, 0.584 mmol, 2.0 equiv.), Pd(PPh₃)₂Cl₂ (41.0 mg, 0.058 mmol, 0.2 equiv.), and CuI (16.7 mg, 0.088 mmol, 0.3 equiv.) as described for the preparation of 2b. The total reaction time was 3 h, but already after 2 h more Pd(PPh₃)₂Cl₂ (20.5 mg, 0.029 mmol, 0.1 equiv.) was added. Flash chromatography [2 cm, 10 ml, diethyl ether/pentane (1:4 \rightarrow 1:2), F 14-16] yielded 43.2 mg (60%). - ¹H NMR (200 MHz): $\delta = 0.18$ [s, Si(CH₃)₃], 2.03 (br. s, OH), 2.42-2.53 (m, $3-H_2$ or $4-H_2$), 2.56-2.68 (m, $3-H_2$ or 4-H₂, 3"-H₂), 3.77 (t, $J_{4",3"} = 6.2$, 4"-H₂), 5.50 (m_c, 1'-H), 6.55 (m_c, 2-H). $- {}^{13}$ C NMR (50 MHz): $\delta = 0.20$ [Si(CH₃)₃], 24.18 (C-3"), 30.52, 31.52 (C-3, C-4), 60.93 (C-4"), 90.86, 102.59 (2 sp C), 98.47 (C-1'), 126.63 (C-1), 149.92 (C-2), 155.26 (C-5); 2 sp-C signals were obscured by the spectral noise. – IR: $\tilde{v} = 3625 \text{ cm}^{-1}$, 2960, 2930, 2850, 2245, 2165, 2115, 1610, 1425, 1250, 1080, 1050, 845, 815.

C₁₅H₂₀OSi Calcd. 244.1283 Found 244.1283 (EI HRMS)

(Z)-1-(5-Hydroxy-1-pentynyl)-5-[3-(trimethylsilyl)-2-propynylidene]-1-cyclopentene (2j) was prepared from 8f (98.1 mg, 0.302 mmol), 4-pentyn-1-ol (56.0 μl, 0.605 mmol, 2.0 equiv.), Pd(PPh₃)₂Cl₂ (42.4 mg, 0.060 mmol, 0.2 equiv.), and CuI (17.3 mg, 0.091 mmol, 0.3 equiv.) as described for the preparation of 2b. The total reaction time was 3 h, but already after 1.5 h more Pd(PPh₃)₂Cl₂ (21.2 mg, 0.030 mmol, 0.1 equiv.) was added. Flash chromatography [2 cm, 10 ml, diethyl ether/pentane (1:3 \rightarrow 3:1), F 16-18] gave 47.6 mg (contaminated with small amounts of an aromatic compound; 57%). - ¹H NMR (200 MHz): $\delta = 0.13$ [s, Si(CH₃)₃], 1.76 (br. s, OH), 1.82 (tt, $J_{4''.5''} \approx J_{4''.3''} \approx 6.4$, $4''-H_2$), 2.40 – 2.52 (m, 3-H₂ or 4- H_2 , 3"- H_2), 2.57 – 2.68 (m, 3- H_2 or 4- H_2), 3.77 (t, $J_{5'',4''} = 6.1$, 5"- H_2), 5.48 (m_c, 1'-H), 6.51 (m_c, 2-H). - ¹³C NMR (50 MHz): $\delta = 0.23$ [Si(CH₃)₃], 16.42, 30.49, 31.21, 31.52 (C-3, C-4, C-3", C-4"), 61.84 (C-5"), 75.39, 93.83, 99.16, 102.56 (sp C), 98.35 (C-1'), 126.94 (C-1), 149.38 (C-2), 155.59 (C-5). - IR: $\tilde{v} = 3625$ cm⁻¹, 3560, 2960, 2930, 2850, 2245, 2165, 2120, 1610, 1430, 1250, 1080, 845, 815.

C₁₆H₂₂OSi Calcd. 258.1440 Found 258.1436 (EI HRMS)

Structure Determination of 3^[16]. - Crystal Data: C₈H₆F₆O₆S₂ (376.2); colorless crystal, size: $0.3 \times 0.3 \times 0.3$ mm; monoclinic; $P2_1/c$; a = 9.612(2), b = 14.335(2), c = 9.814(2) Å; $\beta = 97.530(10)^\circ$; $V = 1340.6(4) \text{ Å}^3$; Z = 4; $\varrho(\text{calcd.}) = 1.864 \text{ Mg/m}^3$; $\mu = 4.568$ mm^{-1} ; F(000) = 752. – Data Collection: Enraf-Nonius CAD4 diffractometer; Cu- K_{α} ($\lambda = 1.54178$ Å); T = 183 K; graphite monochromator; $6.2^{\circ} \le 2\Theta \le 110.0^{\circ}$; ω scan; prescan limit, σ -option: $\sigma = 3.00 \cdot I$, 0.015; prescan speed parameter: 1; maximum scan time: 20.00 s; $\omega = (1.20 + 0.14 \text{ tg}\Theta)^{\circ}$; 3 orientation reflections measured every 250 reflections; $0 \le h \le 10, -15 \le k \le 3, -10$ $\leq l \leq 10$; 2282 reflections collected; 1665 independent reflections $(R_{\rm int} = 5.39\%)$; 1617 observed reflections $[F > 4.0 \sigma(F)]$. - Solution and Refinement: Siemens SHELXTL PLUS (VMS); Direct Methods; full-matrix least squares; minimized quantity: $\sum w(F_0 F_c$)²; extinction correction: $\chi = 0.0031(3)$, where $F^* = F[1 +$ $0.002 \chi F^2/\sin(2\Theta)$]^{-1/4}; hydrogen atoms: refined, common isotropic U; weighting scheme: $w^{-1} = \sigma^2(F) + 0.0000 \cdot F^2$; 219 parameters refined; R = 5.16%, wR = 5.27% (observed data); R = 5.34%, wR = 5.86% (all data); goodness of fit: 6.93; largest, mean Δ/σ : 0.004, 0.001; data-to-parameter ratio: 7.4:1; largest difference peak:

Table 3. Coordinates and equivalent isotropic temperature factors U for 3 (U calculated as one third of the trace of the orthogonal U_{ij} tensor)

	×	у	z	U(eq)
S(1)	2623(1)	-967(1)	-2499(1)	30(1)
S(2)	6663(1)	-2315(1)	-1005(1)	25(1)
F(1)	3505(3)	-339(2)	-4707(3)	64(1)
F(2)	1661(4)	-1163(3)	-5049(4)	107(2)
F(3)	3629(4)	-1816(2)	-4475(3)	85(2)
F(4)	8396(3)	-3566(2)	62(3)	55(1)
F(5)	8795(3)	-3097(2)	-1905(3)	58(1)
F(6)	9343(3)	-2234(2)	-138(3)	64(1)
0(1)	4163(3)	-883(2)	-1810(3)	28(1)
0(2)	2125(3)	-1832(2)	-2110(3)	45(1)
0(3)	1890(3)	-135(2)	-2362(4)	52(1)
0(4)	6829(3)	-1620(2)	-2203(3)	26(1)
0(5)	6480(3)	-1843(2)	209(3)	37(1)
0(6)	5759(3)	-3027(2)	-1571(3)	39(1)
C(1)	7239(4)	-682(3)	-1892(4)	25(1)
C(2)	8522(5)	-377(3)	-1824(5)	37(2)
C(3)	8603(5)	647(3)	-1532(6)	45(2)
C(4)	7065 (5)	943(3)	-1591(5)	33(2)
C(5)	6218(4)	41(2)	-1743(4)	24(1)
C(6)	4854(5)	5(3)	-1728(4)	29(2)
C(7)	2871(6)	-1080(3)	-4301(5)	47(2)
C(8)	8423(5)	-2840(3)	-734(5)	36(2)

 0.44 eÅ^{-3} ; largest difference hole: -0.35 eÅ^{-3} . — Coordinates and temperature factors see Table 3.

CAS Registry Numbers

2a: 137568-96-8 / 2b: 137568-97-9 / 2c: 137568-98-0 / 2d: 137568-99-1 / 2e: 137569-00-7 / 2f: 137569-01-8 / 2g: 137569-02-9 / 2h: 137569-03-0 / 2i: 137569-04-1 / 2j: 137569-05-7 / 3: 134329-52-5 / 4: 1192-54-7 / 6: 134251-47-1 / 8a: 137568-91-3 / 8b: 137568-92-4 / 8c: 137568-93-5 / 8d: 137568-94-6 / 8e: 137568-95-7 / 8f: 128878-94-6 / 8c: 137568-95-7 / 8f: 128878-95-7 / 8f: 128878-95-95-7 / 8f: 128878-95-9 / 8f: 128878-95-88-6 / HC≡CCH₂OTHp: 6089-04-9 / HC≡CCH₂OH: 107-19-7 / HC≡CCH₂CH₂OH: 927-74-2 / HC≡CCH₂CH₂CH₂OH: 5390-04-5 / HC≡CCH₂TMS: 13361-64-3 / HC≡C−TMS: 1066-54-2

[1] N. Ishida, K. Miyazaki, K. M. Kumagai, M. Rikimura, J. An-

tibiot. 1965, 18, 68-76.

[2] Structure: [2a] K. Edo, M. Mizugaki, Y. Koide, H. Seto, K. Furihata, N. Otake, N. Ishida, Tetrahedron Lett. 1985, 26, 331–334. – [2b] A. G. Myers, P. J. Proteau, T. M. Handel, J. Am. Chem. Soc. 1988, 110, 7212–7214.

[3] The modes of action of Neocarzinostatin chromophore and related antibiotics were recently reviewed by I. H. Goldberg, Acc. Chem. Res. 1991, 24, 191–198, and by K. C. Nicolaou, W.-M. Dai, Angew. Chem. 1991, 103, 1453–1481; Angew. Chem., Int.

Fujiwara, A. Kurisaki, M. Hirama, *Tetrahedron Lett.* **1990**, *31*, 4329–4332. – ^[4e] T. Wehlage, A. Krebs, T. Link, *Tetrahedron Lett.* **1990**, *31*, 6625–6628. – ^[4f] J. Suffert, *Tetrahedron Lett.* **1990**, *31*, 7437–7440. – ^[4g] K. Nakatani, K. Arai, N. Hirayama, F. Matsuda, S. Terashima, *Tetrahedron Lett.* 1990, 31, 2323–2326. – [4h] A. G. Myers, P. M. Harrington, E. Y. Kuo, J. Am. Chem. Soc. 1991, 113, 694–695. – [4h] P. Magnus, T. Pitterna, J. Chem. Soc., Chem. Commun. 1991, 541–543. – [4h] K. Fujiwara, H. Sakai, M. Hirama, J. Org. Chem. 1991, 56, 1688–1689. – [4k] T. Doi, T. Takahashi, J. Org. Chem. 1991, 56, 3465–3467. – [4l] M. Hirama, M. Tokuda, K. Fujiwara, Synlett 1991, 651 – 653.

[5] R. Brückner, S. W. Scheuplein, J. Suffert, Tetrahedron Lett. 1991, 32, 1449 - 1452

^[6] J. Suffert, R. Brückner, Tetrahedron Lett. 1991, 32, 1453-1456. ^[7] K. Nakatani, K. Arai, K. Yamada, S. Terashima, *Tetrahedron Lett.* **1991**, 32, 3405–3406.

[8] P. E. Eaton, P. G. Jobe, Synthesis 1983, 796-797.
[9] J. E. McMurry, W. J. Scott, Tetrahedron Lett. 1983, 24, 979 - 982

[10] Further details of the crystal structure investigation are available on request from the Fachinformationszentrum Karlsruhe, Gesellschaft für wissenschaftlich-technische Information mbH, D-7514 Eggenstein-Leopoldshafen 2, on quoting the depository number CSD-320312, the names of the authors, and the journal

[11] S. Cacchi, E. Morera, G. Ortar, Synthesis 1986, 320-322.
[12] Monocoupling product 8b was obtained in a slightly different manner, cf. Experimental.

W. C. Still, M. Kahn, A. Mitra, J. Org. Chem. 1978, 43, 2923 - 2925.

T. W. Greene, Protective Groups in Organic Synthesis, John Wiley & Sons, New York, 1981, p. 21.
 In the ¹³C-NMR spectra of 2, the assignment of C-1 (shielded)

vs. C-5 (deshielded) was performed in analogy to Edo's values for $1^{[2a]}$ ($\delta_{C-1}=129.8,\,\delta_{C-5}=160.2$). In the H-decoupled ¹³C-NMR spectra of 2, C-2 (tertiary) and C-5 (quaternary) were distinguished by the more intense resonance of the former.

(16) N. Walker, D. Stuart, Acta Crystallogr., Sect. A: Found. Crystallogr. 1983, A39, 158 – 166; A. L. Spek, PLATON 88, Program for Geometrical Analysis of Crystal Structures, Utrecht, 1988; E. Keller, SCHAKAL-88B, A FORTRAN Program for the Graphic Representation of Molecular and Crystallographic Models, Univ. of Freiburg, 1988.

[351/91]